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"The signature undertaking of the Twenty-Second Edition was clarifying the QC practices necessary to perform the methods in this manual. Section in Part 1000 were rewritten, and detailed QC sections were added in Parts 2000 through 7000. These changes are a direct and necessary result of the mandate to stay abreast of regulatory requirements and a policy intended to clarify the QC steps considered to be an integral part of each test method. Additional QC steps were added to almost half of the sections."--Pref. p. iv.

Physicochemical and Environmental Plant Physiology provides an understanding of various areas of plant physiology in particular and physiology in general. Elementary chemistry, physics, and mathematics are used to explain and develop concepts. The first three chapters of the book describe water relations and ion transport for plant cells. The next three chapters cover the properties of light and its absorption; the features of chlorophyll and the accessory pigments for photosynthesis that allow plants to convert radiant energy from the sun into chemical energy; and how much energy is actually carried by the compounds ATP and NADPH. The last three chapters consider the various forms in which energy and matter enter and leave a plant as it interacts with its environment. These include the physical quantities involved in energy budget analysis; the resistances affecting the movement of both water vapor and carbon dioxide in leaves; and the movement of water from the soil through the plant to the atmosphere.

This text is the successor volume to Biophysical Plant Physiology and Ecology (W.H. Freeman, 1983). The content has been extensively updated based on the growing quantity and quality of plant research, including cell growth and water relations, membrane channels, mechanisms of active transport, and the bioenergetics of chloroplasts and mitochondria. One-third of the figures are new or modified, over 190 new references are incorporated, the appendixes on constants and conversion factors have doubled the number of entries, and the solutions to problems are given for the first time. Many other changes have emanated from the best laboratory for any book, the classroom. · Covers water relations and ion transport for plant cells; diffusion, chemical potential gradients, solute movement in and out of plant cells · Covers interconnection of various energy forms; light, chlorophyll and accessory photosynthesis pigments, ATP and NADPH · Covers forms in which energy and matter enter and leave a plant; energy budget analysis, water vapor and carbon dioxide, water movement from soil to plant to atmosphere

Water is basic to terrestrial life, and its distribution has controlled the growth and spread of human civilization. The importance of water to modern industrial processes, urban planning, and agricultural development is hard to overestimate. With these compelling motivations, it is

natural that more technical and scientific study should have been devoted to this one substance than to any other. Research on water and its solutions has exhibited a marked expansion during the last decade. In significant degree, this has resulted from the availability of new experimental tools and techniques, and of dramatic advances in computing science. This combination, in skilled hands, promises eventually to explain the unusual properties of water and aqueous solutions in unequivocal molecular terms. Like wise, one now has reasonable hope that the active role that water plays in biochemical processes will be revealed and explained quantitatively at the molecular level. Owing to the widespread scholarly interest in aqueous science, it is clear that guides to the overwhelming literature on the subject are valuable. They serve ideally to indicate what is known and what is not, which areas harbor controversies, and what types of research attacks seem most fruitful (in answering more questions than they raise!). Whatever time and resources need to be spent in preparing comprehensive bibliographies should be quickly offset in the total scientific community by the efficiencies generated.

In addition to detailed instructions for sampling and immediate analysis, the book provides a concise presentation of both the theoretical background and data evaluation. The analytical methods thus presented can just as easily be applied using simple equipment as well as in the modern laboratory. The book is a bench-top laboratory manual and as such can be used for instruction in laboratory staff training programs. It treats the analysis of organic and inorganic compounds while also dealing with microbiological problems associated with the guidelines for waste, surface and ground water, as well as drinking water quality.

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