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Chemistry test: 1) Each a-particle has the mass: a) Equal to that of hydrogen. b) Nearly four times that of hydrogen atom. c) Half of the hydrogen atom. d) None

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of these. ANSWER : Nearly four times that of hydrogen atom. 2) Which among the following has the highest penetrating power? a) γ - rays b) β - rays

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Q. Barium-122 has a half-life of 2 minutes. A fresh sample weighing 80 g was obtained. If it takes 10

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minutes to set up an experiment using barium-122, how much barium-122 will be left when the experiment begins?

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answer choices. A plutonium atom is used to start a chain reaction that detonates a nuclear weapon. A uranium atom is split

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apart into lighter elements. Two hydrogen atoms are combined to form a helium atom. Two hydrogen atoms bond with an oxygen atom to form a water molecule. Tags:

Nuclear Chemistry Practice

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Test Quiz - Quizizz

Nuclear Half Lives and
Radioactive Decay Math p7
Answer Key p11. Key
Equations Given for Test:
 $E_{\text{cell}} = E_{\text{reduction}} + E_{\text{oxidation}}$
 $\Delta G = -96.5nE_{\text{cell}}$ (ΔG in kJ)

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$E_{\text{cell}} = E^{\circ} - [0.0592/n] \log Q$
 $\log K = nE^{\circ}/0.0592$ Mol e^{-} =
[A • time (sec)/96,500] time
(sec) = mol e^{-} •
96,500/current (in A) $t =$
($t_{1/2} / 0.693$) $\ln (A)$.

Radioactivity and Balancing

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Nuclear Reactions: Balancing

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... Identify the missing
particle in the following
nuclear reaction: ${}_{99}^{43}\text{Tc} ?$

 + ${}_{-1}^0\text{e} ?$ ${}_{99}^{44}\text{Ru} ?$

${}_{99}^{43}\text{Tc} ?$ ${}_{99}^{42}\text{Mo} ?$ ${}_{99}^{43}$

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Es; What is the effect of this decay? ${}_{99}^{43}\text{Tc} \rightarrow {}_{99}^{44}\text{Ru} + {}_{0}^{-1}\text{e}$? a neutron is converted to a proton ?

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Regents review Nuclear

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Chemistry Base your answers to questions 52 through 54 on the information below.

Nuclear radiation is harmful to living cells, particularly to fast-growing cells, such as cancer cells and blood cells. An external

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beam of the radiation
emitted from a radioisotope

Regents review Nuclear Chemistry 2011-2012

Identify the following based
on the mass number and
atomic number: 0?. 0 A)

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alpha particle B) beta particle C) positron particle D) gamma ray E) neutron. 7. Gamma rays require the heaviest shielding of all the common types of nuclear radiation because gamma rays have the

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A) highest energy. B) most intense color.

Nuclear Chemistry Practice Problems

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asked in csir nuclear
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solutions chemistry
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Nuclear Chemistry (Radioactivity) Exams and Problem ...

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done! 1. The smallest
particle of an element that

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retains the properties of
that element. 2. 3.

Radioactive, has an unstable
nucleus and undergoes
radioactive decay. 4.

Positively charged particle
emitted from certain
radioactive nuclei, consists

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of two protons and two
neutrons.

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CHAPTER 22 TEST Nuclear
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CHOICE On the line at the left of each statement, write the letter of the choice that best completes the statement or answers the question. After converting units, the nuclear mass defect is equivalent to the

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a. atomic mass b.
electrostatic force c.
energy of chemical reaction

San Ramon Valley High School

These processes takes place
in the nucleus and are
called nuclear reactions.

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This quiz is over simple nuclear chemistry. Refer to the diagram and images to aid in answering the questions. Using what you have learned, select the best answer to the choices.
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Quizzes : Topic: Nuclear
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Chemistry Study Guide: Test
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